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State of Matter

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Pure Substances and
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States of Matter :

Solid Liquid Gas

Lewis Diagrams Made
Easy: How to Draw

Lewis Dot Structures

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States of Matter

/u0026 Changing

State #20

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and 10 Types of

~~Matter: Elements,~~

~~Compounds, and~~

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Chemical Change

cation. anion. ionic
bond. the force that
holds two atoms

together; may form
by the attra....

positively charged
ion; forms when
valence electrons are
remov.... negatively
charged ion; forms
when valence
electrons are
added....

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CHAPTER 8
SOLUTIONS MANUAL

Covalent

Bonding Covalent

Bonding Solutions

Manual Chemistry:

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• Chapter 8 121

Section 8.1 The

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Covalent Bond pages

240–247 Practice

Problems page 244

Answer Key

Draw the Lewis

structure for each

molecule. 1. PH_3

H_2O H_2S

respectively, for

single, double, and

triple $\text{P} \equiv \text{P}$ 2. H_2S

H_2O H_2S ...

Chemistry Matter

And Change Chapter

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~~8 Worksheet Answers~~

CHAPTER 8 Table Of

Contents • Apply

the octet rule to

atoms that form

covalent bonds.

chemical bond: the

force that holds two

atoms together •

Describe the

formation of single,

double, and triple

covalent bonds. •

Contrast sigma and pi

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bonds. • Relate the strength of a covalent bond to its bond length and bond dissociation energy.

~~CMC Chapter 08—
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8...~~

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Chapter 8. diatomic.
single covalent bond.

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Chapter 8
Answer Key

double covalent
bond. seven diatomic
molecules

(BrINClHO_F) two

atoms of the same

element. 1 bond- 2e.

2 pairs of e- are

shared. hydrogen,

oxygen, nitrogen,

flourine, chlorine,

bromine, iodine.

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8 44 . Name Date
CHAPTER FOR Class
Section 8.2 continued
9 What is the
relationship between
lattice energy and the
strength of the
attractive force
holding ions in place?

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- a. The more positive the lattice energy is, the greater the force.
- b. The more negative the lattice energy is, the greater the force.

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solutions reorient

your old paradigms.

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make today the first

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Fulfillment) today.

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CHEMISTRY Matter
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8: Covalent Bonding

This graph
summarizes the
range of chemical
bonds between two
atoms. A. hydrogen

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bond B. covalent bond C. ionic bond D.

dipole bond 86

Covalent Bonding

CHAPTER8 Chapter

Assessment Give the

correct name for the

molecule... 20

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8 Assessment

Answers

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Solutions Manual

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Chapter 8 123 34.

Apply Using the
system of rules for
naming binary
molecular

compounds, describe
how you would name
the molecule N_2O_4 .

There are two atoms
of nitrogen; use the
prefix di- with the

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name nitrogen. There are four atoms of oxygen, so use the prefix tetra—the root of

Covalent

Bonding Covalent

Bonding

8. In a crystal lattice of an ionic compound, a. ions of a given charge are clustered together,

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far from ions of the
opposite charge. ions
are surrounded by
ions of the opposite
charge. c. a sea of
electrons surrounds
the ions. d. neutral
molecules are
present. Chemistry:
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8.5: Chapter

Summary Last

updated; ... the

meanings of the

following bold terms

in the following

summary and ask

yourself how they

relate to the topics in

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the chapter. A phase

is a certain form of matter that has the same physical

properties

throughout. Three

phases are common:

the solid, the liquid,

and the gas phase. ...

Because the ...

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4c5ba27e0d3

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