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Introduction to
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Hydrogen evolution
and Oxygen

Absorption

Mechanism of wet
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Physical Chemistry
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construction of cell |
electrochemical cell |
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noc18-mm14 Lecture

01-Introduction to

corrosion-1

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~~Introduction To~~

~~Electrochemistry 01~~

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~~Electrochemical
mechanism of rusting
of iron (evolution of
hydrogen and
absorption of oxygen~~

} Galvanic corrosion

Galvanic Corrosion |

Forms of Corrosion

Standard Hydrogen

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Corrosion Galvanic

Cells (Voltaic Cells)

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theory of corrosion

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Cathodic Protection
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Sacrificial Anode
method | Impressed
Current Method 25.

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FuseSchool How does
galvanic corrosion
and pitting corrosion

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occur? (Bimetallic
corrosion)

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Combustion For TNEB

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corrosion,

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ELECTROCHEMISTRY
-27 || CORROSION ||
RUSTING OF IRON

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chap 3~~

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cal/Electrolytic
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Revision Notes on

Redox Reactions and

Electrochemistry:

Electrochemistry is

the branch of

chemistry which

deals with the

chemical changes

caused in the matter

by passage of electric

current and

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conversion of
chemical energy into
electrical energy and
vice versa.

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Molar Conductivity),

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Variation of specific
conductivity with
dilution, etc

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Notes - 2020 | SW

An atom is made up
of sub atomic
particles like electrons,
protons and neutrons
etc. Electro chemistry
is a branch of
chemistry which

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deals with the
transformation of
electrical energy into
chemical energy or
chemical into
electrical energy.

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MODULE I

ELECTRODE

POTENTIAL & CELLS

Electrochemistry: It is
a branch of chemistry
which deals with the
study of

transformation of
chemical energy into
electrical energy and

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rochemistry

vice versa ”
Electrochemical cell
and Classification
with examples.

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Energy Systems |
Chemical ...~~

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Electrochemistry

Electrochemistry is
that branch of
chemistry which
deals with the study
of production of
electricity from
energy released
during spontaneous

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chemical reactions
and the use of
electrical energy to
bring about non-
spontaneous
chemical
transformations.
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